# Inversion of $3_{10}$-Helix Screw Sense in a ( $D-\alpha$ Me)Leu Homotetrapeptide Induced by a Guest $\mathrm{D}-(\alpha \mathrm{Me})$ Val Residue 

FERNANDO FORMAGGIO', MARCO CRISMA', CLAUDIO TONIOLO', ETTORE BENEDEITI ${ }^{2}$, BENEDETTO DI BLASIO ${ }^{2}$, MICHELE SAVIANO ${ }^{2}$, STEFANIA GALDIERO ${ }^{2}$, JOHAN KAMPHUIS ${ }^{3}$ and ANTONELLO SANTINI ${ }^{4}$<br>' Biopolymer Research Centre, CNR, Department of Organic Chemistry, University of Padova, Italy<br>${ }^{2}$ Biocrystallography Centre, CNR, Department of Chemistry, University of Naples 'Federico II', Naples, Italy<br>${ }^{3}$ DSM Research, Bio-organic Chemistry Section, Geleen, The Netherlands<br>${ }^{4}$ Department of Food Science, University of Naples 'Federico II', Naples, Italy


#### Abstract

The terminally blocked tetrapeptide $\mathrm{pBrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{2}-\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Val}-\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}-\mathrm{OtBu}$ is folded in the crystal state in a left-handed $3_{10}$-helical structure stabilized by two consecutive $1 \leftarrow 4 \mathrm{C}=\mathrm{O} \cdots \mathrm{H}-\mathrm{N}$ intramolecular H-bonds, as determined by X-ray diffraction analysis. A CD study strongly supports the view that this conformation is also that largely prevailing in MeOH solution. A comparison with the published conformation of $p \mathrm{BrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{4}$-OtBu indicates that incorporation of a single internal $\beta$-branched $(\alpha \mathrm{Me}) \mathrm{Val}$ guest residue into the host homo-tetrapeptide from the $\gamma$-branched ( $\alpha \mathrm{Me}$ )Leu residue is responsible for a dramatic structural perturbation, i.e. an inversion of the $3_{10}$ screw sense from right to left-handed.


Keywords: ( $\alpha \mathrm{Me}$ ) amino acids; CD spectroscopy; $3_{10}$-helix; peptide 3D-structure; X-ray structure


#### Abstract

Abbreviations $p \mathrm{BrBz}$ para-bromobenzoyl; ( $\alpha \mathrm{Me}$ )Leu, $\mathrm{C}^{\alpha}$-methyl leucine; ( $\alpha \mathrm{Me}$ )Val, $\mathrm{C}^{\alpha}$-methyl valine; $\mathrm{Abu}, \alpha$-aminobutyric acid; Deg, $\mathrm{C}^{\alpha, \alpha}$-diethylglycine; OtBu, tert-butoxy; EDC, $N$-ethyl- $\boldsymbol{N}^{\prime}$-[3-dimethylamino) propyl] carbodiimide.


## INTRODUCTION

In the late 1970s Goodman and co-workers [1, 2] pioneered the approach of synthetically inserting a single guest protein amino acid into selected positions of a well-characterized, monodispersed homo-

[^0]oligopeptide chain with the aim of gaining useful information into the factors influencing the conformational stability of peptides. We have recently extended this methodology to the field of $\mathrm{C}^{\alpha^{, \alpha}}$ disubstituted glycines which, in addition to the classical $\alpha$-helix, are known to prefer less common secondary structures, such as the $3_{10}$-helix and the fully extended conformation (or $2.0_{5}$ helix) [3, 4]. More specifically, we have been able to demonstrate that a modest point defect, represented by the incorporation of an Abu guest residue interrupting the side-chain uniformity of the host (Deg) ${ }_{5}$ homopeptide, while altering only marginally the conformation in a solvent of low polarity, is responsible for a dramatic perturbation of the crystal state structure, from the flat fully extended conformation of the host pentapeptide to the $3_{10}$-helix of the host/guest analogue [5].

Recent works from our laboratories have shown that the $3_{10}$-helix is preferentially adopted by peptides rich in $\mathrm{C}^{\alpha}$-methylated chiral amino acids [4]. Intriguing experimental findings on the impact of $\mathrm{C}^{\alpha}$
chirality on helix screw sense have been reported, which demonstrate that $\mathrm{C}^{\alpha}$-methylated amino acids with a linear or a $\beta$-branched side chain exhibit a normal relationship (i.e. the same as that shown by protein amino acids: L-amino acids give right-handed helices), whereas the relationship characterizing $\mathrm{C}^{\alpha}$ methylated amino acids with a $\gamma$-branched side chain is opposite.

As a part of our ongoing study aiming at establishing the implications of point defects (e.g. side-chain irregularities) on the secondary structure of $\mathrm{C}^{\alpha, \alpha}$ disubstituted glycine-rich homo-peptides, we describe in this paper the experimental results obtained using X-ray diffraction and CD spectroscopy on the terminally blocked tetrapeptide $p \mathrm{BrBz}-[\mathrm{D}-$ $(\alpha \mathrm{Me}) \mathrm{Leu}]_{2}-\mathrm{D}-(\alpha \mathrm{Me})$ Val-D-( $\alpha \mathrm{Me}$ )Leu-OtBu. In comparison with the ( $\alpha \mathrm{Me}$ )Leu residue of the host homotetrapeptide $p \mathrm{BrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{4}-\mathrm{OtBu}$, the single ( $\alpha \mathrm{Me}$ )Val residue of the host/guest peptide has one side-chain carbon atom less, i.e. it is a $\beta$-branched, instead of a $\gamma$-branched, amino acid.

## MATERIALS AND METHODS

## Synthesis and Characterization of Peptides

Melting points were determined using a Leitz (Wetzlar, Germany) model Laborlux 12 apparatus and are not corrected. Optical rotations were measured using a Perkin-Elmer (Norwalk, CT) model 241 polarimeter equipped with a Haake (Karlsruhe, Germany) model D thermostat. Thin-layer chromatography was performed on Merck (Darmstadt, Germany) Kieselgel $60 \mathrm{~F}_{254}$ precoated plates using the following solvent systems: $1\left(\mathrm{CHCl}_{3}-\mathrm{EtOH}, 9: 1\right), 2(n B u O H-A c O H-$ $\mathrm{H}_{2} \mathrm{O}, 3: 1: 1$ ); 3 (toluene-EtOH, 7:1). The chromatograms were examined by UV fluorescence or developed by chlorine-starch-potassium iodide or ninhydrin chromatic reaction as appropriate. All the compounds were obtained in a chromatographically homogeneous state. The solid-state IR absorption spectra ( KBr disk technique) were recorded with a Perkin-Elmer model 580B spectrophotometer equipped with a Perkin-Elmer model 3600 data station. The ${ }^{1} \mathrm{H}$-NMR spectra were recorded with a Bruker model AC 250 spectrometer. Measurements were carried out in deuterochloroform ( $99.96 \% d$, Merck) with tetramethylsilane as the internal standard.
pBrBz-(D-( $\alpha \mathrm{Me}) \mathrm{Leu})_{2}-\mathrm{D}(-\alpha \mathrm{Me}) \mathrm{Val}-\mathrm{OtBu}$. The $5(4 \mathrm{H})-$ oxazolone from $p \mathrm{BrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{2}-\mathrm{OH}[6](0.14 \mathrm{~g}$, 0.32 mmol ) and $\mathrm{H}-\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Val}-\mathrm{OtBu}$ (obtained from
$\mathrm{Pd}-\mathrm{C}$ catalysed hydrogenolysis of the corresponding Z-protected amino acid ester [7] ( $0.40 \mathrm{~g}, 1.24 \mathrm{mmol}$ ) were refluxed in $\mathrm{CH}_{3} \mathrm{CN}$ for 6 h . The solvent was removed under reduced pressure, the organic phase washed with $10 \% \mathrm{KHSO}_{4}$, water, $5 \% \mathrm{NaHCO}_{3}$ and water, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated to dryness. The product was purified by flash-chromatography by eluting the column with a $1: 5$ isocratic mixture of AcOEt : petroleum ether. Yield 78\%; m.p. 194-195 ${ }^{\circ} \mathrm{C}$ (from AcOEt/petroleum ether); $\quad R_{\mathrm{F} 1}=0.95$, $R_{R_{\mathrm{r} 2}}=0.95, R_{\mathrm{F} 3}=0.60 ;[\alpha]_{\mathrm{D}}^{20}=6.4^{\circ}(c=0.5, \mathrm{MeOH})$, $\left[\alpha{ }_{436}^{20}=15.9^{\circ}(\mathrm{c}=0.5, \mathrm{MeOH})\right.$; IR (KBr): 3403, 3363, 1715, 1672, $1651 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right.$, 50 mM ): $\delta 7.92$ (s, $1 \mathrm{H}, \mathrm{NH}), 7.31$ (m, $5 \mathrm{H}, 4 \mathrm{pBrBz}-$ phenyl CH and 1 NH ), $6.84(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 2.70(\mathrm{~m}, 2 \mathrm{H}$, Leu $\beta \mathrm{CH}_{2}$ ), $2.45(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Val} \beta \mathrm{CH}), 1.72(\mathrm{~s}, 3 \mathrm{H}, \mathrm{Val}$ $\beta \mathrm{CH}_{3}$ ), 1.62 and 1.61 ( $2 \mathrm{~s}, 6 \mathrm{H}$, 2Leu $\beta \mathrm{CH}_{3}$ ), 1.56 (m, 4 H , Leu $\beta \mathrm{CH}_{2}$, 2Leu $\gamma \mathrm{CH}$ ), 1.50 ( $\mathrm{s}, 9 \mathrm{H}, \mathrm{OtBu} \mathrm{CH}_{3}$ ), 1.01 (d, 3 H , Val $\gamma \mathrm{CH}_{3}$ ), $0.86\left(\mathrm{~m}, 15 \mathrm{H}\right.$, Val $\gamma \mathrm{CH}_{3}$ and 4Leu $\delta \mathrm{CH}_{3}$ ).
$\mathrm{pBrBz}-(\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{LeU})_{2}-\mathrm{D}(-\alpha \mathrm{Me}) \mathrm{Val}-\mathrm{OH}$. This compound was prepared by treatment of $\mathrm{pBrBz}-[\mathrm{D}-$ $(\alpha \mathrm{Me}) \mathrm{Leu}]_{2}$-D- $(\alpha \mathrm{Me}) \mathrm{Val}-\mathrm{OtBu}(0.13 \mathrm{~g}, 0.21 \mathrm{mmol})$ with a 2:1 TFA: $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ mixture for 90 min at room temperature. The solvent mixture was removed in vacuo. The resulting oil, treated several times with diethyl ether, afforded the solid title compound. Yield $90 \%$; m.p. $\quad 196-197^{\circ} \mathrm{C} ; \quad R_{\mathrm{F} 1}=0.45, \quad R_{\mathrm{F} 2}=0.95$; $R_{\mathrm{F} 3}=0.25 ; \quad[\alpha]_{\mathrm{D}}^{20}=-10.4^{\circ} \quad(c=0.5, \quad \mathrm{MeOH})$, $[\alpha]_{436}^{20}=-20.9^{\circ}(c=0.5, \mathrm{MeOH}) ; \operatorname{IR}(\mathrm{KBr}): 3321$, 1735, $1647 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 50 \mathrm{~mm}\right): \delta 7.62$ ( $\mathrm{m}, 5 \mathrm{H}, 4 \mathrm{pBrBz}-\mathrm{phenyl} \mathrm{CH}$ and 1 NH ), 7.28 (s, 1 H , $\mathrm{NH}), 6.84(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 6.67(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 2.54(\mathrm{~m}, 1 \mathrm{H}$, Val $\beta \mathrm{CH}_{2}$ ), 1.85 (m, $6 \mathrm{H}, 2 \mathrm{Leu} \beta \mathrm{CH}_{2}$ and 2Leu $\gamma \mathrm{CH}$ ), $1.64,1.58$ and $1.54\left(3 \mathrm{~s}, 9 \mathrm{H}, 2 \mathrm{Leu} \beta \mathrm{CH}_{3}\right.$ and Val $\beta \mathrm{CH}_{3}$ ), 1.08 (d, 3H, Val $\gamma \mathrm{CH}_{3}$ ), $0.99\left(\mathrm{~m}, 9 \mathrm{H}\right.$, Val $\gamma \mathrm{CH}_{3}$ and $2 \mathrm{Leu} \delta \mathrm{CH}_{3}$ ), $0.83\left(2 \mathrm{~d}, 6 \mathrm{H}\right.$, 2Leu $\left.\delta \mathrm{CH}_{3}\right)$.

5(4H)-Oxazolone from $\mathrm{pBrBz-(D-( } \mathrm{\alpha Me)Leu})_{2}$-D$(\alpha \mathrm{Me}) \mathrm{Val}-\mathrm{OH}$. This compound was prepared from $p \mathrm{BrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{2}-\mathrm{D}-(\alpha \mathrm{Me})$ Val-OH $(0.094 \mathrm{~g}, 0.165$ $\mathrm{mmol})$ and EDC•HCl ( $0.035 \mathrm{~g}, 0.182 \mathrm{mmol}$ ) in a $1: 1$ $\mathrm{CH}_{3} \mathrm{CN}$ : AcOEt mixture. After stirring for 1 h at room temperature, the solvent was removed under reduced pressure, the residue dissolved in AcOEt and the organic phase washed with $10 \% \mathrm{KHSO}_{4}$, water, $5 \% \mathrm{NaHCO}_{3}$ and water, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated to dryness. Oil. Yield $93 \%$. $R_{F 1}=0.95, \quad R_{F 2}=0.95, \quad R_{F 3}=0.75 ; \quad[\alpha]_{\mathrm{D}}^{20}=-13.2^{\circ}$ ( $c=0.5$, AcOEt), $[\alpha]_{436}^{20}=-28.2^{\circ}$ ( $c=0.5$, AcOEt); IR ( KBr ): 3375, $1827,1655 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right.$, $10 \mathrm{~mm}): \delta 7.84(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 7.70$ and $7.56(\mathrm{~m}, 4 \mathrm{H}$, $p \mathrm{BrBz}-$ phenyl CH ), 7.29 (s, $1 \mathrm{H}, \mathrm{NH}$ ), $2.84(2 \mathrm{~d}, 1 \mathrm{H}$, Leu $\beta \mathrm{CH}_{2}$ ), $2.49\left(2 \mathrm{~d}, 1 \mathrm{H}\right.$, Leu $\left.\beta \mathrm{CH}_{2}\right), 2.05(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Val}$
$\beta \mathrm{CH}$ ), 1.75 and $1.72\left(2 \mathrm{~s}, 6 \mathrm{H}, 2 \mathrm{Leu} \beta \mathrm{CH}_{3}\right), 1.61$ ( m , 4 H , Leu $\beta \mathrm{CH}_{2}$ and 2Leu $\gamma \mathrm{CH}$ ), 1.41 (s, $3 \mathrm{H}, \mathrm{Val} \beta \mathrm{CH}_{3}$ ), $0.90\left(\mathrm{~m}, 18 \mathrm{H}, 4 \mathrm{Leu} \delta \mathrm{CH}_{3}\right.$ and $\left.2 \mathrm{Val} \gamma \mathrm{CH}_{3}\right)$.

## pBrBz-(D-( $\alpha \mathrm{Me}) \mathrm{LeU})_{2}-\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Val}-\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{LeU}-$

 OfBu. This compound was prepared from the $5(4 \mathrm{H})-$ oxazolone from $\mathrm{pBrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{2}-\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Val}-\mathrm{OH}$ ( $0.075 \mathrm{~g}, 0.14 \mathrm{mmol}$ ) and $\mathrm{H}-\mathrm{d}-(\alpha \mathrm{Me}) L e u-\mathrm{OtBu}$ [obtained from Pd-ccatalysed hydrogenolysis of the corresponding Z-protected amino acid ester [6] ( $0.30 \mathrm{~g}, 0.90 \mathrm{mmol}$ )] according to the procedure described above for $p \mathrm{BrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{2}$-D- $(\alpha \mathrm{Me}) \mathrm{Val}-$ OtBu (refluxed in $\mathrm{CH}_{3} \mathrm{CN}$ for 24 h ). The product was purified by flash-chromatography by eluting the column with a $1: 6$ to $1: 4$ step-gradient mixture of AcOEt: petroleum ether. Yield $66 \%$; m.p. 192$193{ }^{\circ} \mathrm{C} ; \quad R_{\mathrm{F} 1}=0.95, \quad R_{\mathrm{F} 2}=0.95, \quad R_{\mathrm{F} 3}=0.60$; $[\alpha]_{\mathrm{D}}^{20}=-16.2^{\circ} \quad(\mathrm{c}=0.5, \quad \mathrm{MeOH}) ; \quad \mathrm{IR}(\mathrm{KBr}): 3412$, 1727, $1659 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$-NMR ( $\mathrm{CDCl}_{3}, 10 \mathrm{~mm}$ ): $\delta 7.89$ (s, 1H, NH), 7.70 and 7.55 ( $2 \mathrm{~m}, 5 \mathrm{H}, 4 \mathrm{pBrBz}$-phenyl CH and 1 NH ), $7.46(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 7.06(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 2.60$ ( $\mathrm{m}, 3 \mathrm{H}$, Leu $\beta \mathrm{CH}_{2}$ and Val $\gamma \mathrm{CH}$ ), 1.65 ( $\mathrm{m}, 7 \mathrm{H}$, 2Leu $\beta \mathrm{CH}_{2}$ and $\left.3 \mathrm{Leu} \gamma \mathrm{CH}\right), 1.74,1.72,1.60$ and $1.55(4 \mathrm{~s}$, 12 H , 3Leu $\beta \mathrm{CH}_{3}$ and Val $\beta \mathrm{CH}_{3}$ ), 1.50 (s, $9 \mathrm{H}, \mathrm{OtBu}$ $\mathrm{CH}_{3}$ ), 1.04 (d, 3 H , Val $\gamma \mathrm{CH}_{3}$ ), $0.92(\mathrm{~m}, 21 \mathrm{H}, 6 \mathrm{Leu}$ $\delta \mathrm{CH}_{3}$ and $\mathrm{Val} \gamma \mathrm{CH}_{3}$ ).
## X-ray Diffraction

Colourless single crystals were obtained by slow evaporation from a methanol solution. A crystal with dimensions $0.2 \times 0.3 \times 0.4 \mathrm{~mm}$ was used for unit cell determination and data collection on a CAD4Turbo Enraf Nonius automated diffractometer using graphite monochromated $\mathrm{CuK} \alpha$ radiation. The crystals are tetragonal, space group $\mathrm{P}_{3}$ with $a=b=19.0574(2) \AA, c=11.3009(3) \AA$ and $Z=4$. A total of 4325 unique reflections were measured at room temperature in the range $0<2 \theta<140^{\circ}, 3669$ of which were classified as observed having $I>3 \sigma(I)$ and used for structure determination and refinement. The structure was solved using the SIR 92 package [8]. The best E map revealed most of the non-H atoms. Remaining ones were recovered from subsequent difference Fourier maps. H atoms were in part located on successive difference Fourier maps, and in part calculated in their stereochemically expected positions. Refinement of the structure was performed by a full matrix least-squares procedure minimizing the quantity $\Sigma w\left(F_{0}^{2}-F_{c}^{2}\right)^{2}$, with $\omega=1 / \sigma\left(F_{\mathrm{o}}^{2}\right)$. All non-H atoms were refined anisotropically. H atoms were introduced in the calculations with isotropic thermal factors equal to the Beq of the

Table 1 Positional Parameters and Their Estimated Standard Deviations for $p \mathrm{BrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) L e u]_{2}-\mathrm{D}-$ ( $\alpha \mathrm{Me}$ )Val-D-( $\alpha \mathrm{Me}$ )Leu-OtBu

| Atom | $x$ | $y$ | $z$ | $B\left(\AA^{2}\right)$ |
| :---: | :---: | :---: | :---: | :---: |
| Br | 0.96037(3) | 0.83814(5) | 0.293 | 7.32(2) |
| C(1) | 0.7547(2) | 0.8293(3) | 0.5129(4) | 3.7(1) |
| $\mathrm{C}(2)$ | 0.8099(3) | 0.7861 (3) | $0.5397(5)$ | 4.4(1) |
| $\mathrm{C}(3)$ | $0.8711(3)$ | 0.7869(3) | $0.4725(5)$ | 5.1(1) |
| C(4) | $0.8768(3)$ | 0.8347(3) | 0.3819(5) | 4.6(1) |
| C(5) | 0.8217(3) | 0.8790(3) | 0.3530(5) | 5.1(1) |
| C(6) | $0.7604(3)$ | 0.8753(3) | 0.4186(5) | 4.8(1) |
| $\mathrm{C}_{0}^{\prime}$ | 0.6879(3) | 0.8306(3) | $0.5811(4)$ | 3.8(1) |
| $\mathrm{O}_{0}$ | $0.6337(2)$ | 0.8556(2) | $0.5386(4)$ | 4.74(8) |
| $\mathrm{N}_{1}$ | 0.6889(2) | 0.8042(2) | 0.6935(4) | 3.73(8) |
| $\mathrm{C}_{1}^{\text {a }}$ | 0.6299(2) | 0.8156(2] | $0.7743(4)$ | 3.43(9) |
| $\mathrm{C}_{1}^{\beta 1}$ | $0.6216(3)$ | $0.8931(3)$ | 0.7992(6) | 4.7(1) |
| $\mathrm{C}_{1}^{\beta}$ | $0.6458(3)$ | 0.7726(3) | $0.8868(5)$ | 3.9(1) |
| $\mathrm{C}_{1}^{\gamma}$ | 0.5922(3) | 0.7796(3) | $0.9883(5)$ | 4.4(1) |
| $\mathrm{C}_{1}^{\text {61 }}$ | $0.5754(4)$ | 0.7100(4) | $1.0384(7)$ | 8.1(2) |
| $\mathrm{C}_{1}^{\text {¢2 }}$ | 0.6173(4) | 0.8283(4) | $1.0848(6)$ | 7.9(2) |
| $\mathrm{C}_{1}^{\prime}$ | 0.5620(2) | 0.7860(3) | $0.7198(4)$ | 3.6(1) |
| $\mathrm{O}_{1}$ | 0.5049(2) | 0.8136(2) | $0.7398(3)$ | 4.43(8) |
| $\mathrm{N}_{2}$ | 0.5691(2) | 0.7279(2) | 0.6540(4) | 3.88(9) |
| $\mathrm{C}_{2}^{\alpha}$ | 0.5106(3) | $0.6861(3)$ | 0.6090(5) | 3.9(1) |
| $\mathrm{C}_{2}^{\text {ف1 }}$ | $0.4716(3)$ | $0.6537(3)$ | 0.7126(6) | 5.5(1) |
| $\mathrm{C}_{2}{ }^{\text {a }}$ | $0.5401(3)$ | $0.6261(3)$ | 0.5309(6) | 4.9(1) |
| $\mathrm{C}_{2}^{\gamma}$ | 0.5660 (3) | 0.6408(3) | 0.4063(6) | 5.0(1) |
| $\mathrm{C}_{1}{ }^{1}$ | 0.5817(4) | $0.5711(4)$ | $0.3448(7)$ | 7.9(2) |
| $\mathrm{C}_{2}{ }^{2}$ | 0.6300(3) | 0.6884(4) | 0.4010(6) | 6.3(2) |
| $\mathrm{C}_{2}^{\prime}$ | 0.4589(2) | 0.7284(2) | 0.5332(5) | 3.9(1) |
| $\mathrm{O}_{2}$ | 0.4000(2) | 0.7041 (2) | 0.5142(4) | 5.24(9) |
| $\mathrm{N}_{3}$ | 0.4817(2) | 0.7896(2) | $0.4893(4)$ | 3.66(8) |
| $\mathrm{C}_{3}^{\alpha}$ | 0.4415(3) | 0.8358(3) | $0.41104(4)$ | 3.8(1) |
| $\mathrm{C}_{3}^{31}$ | 0.4165(3) | 0.7955(3) | $0.3031(5)$ | 5.8(1) |
| $\mathrm{C}_{3}^{\text {3 }}$ | 0.4919(3) | 0.8970 3 ) | $0.3763(5)$ | 4.5(1) |
| $\mathrm{C}_{3}^{71}$ | $0.5051(3)$ | 0.9840(3) | 0.4745(6) | 5.1(1) |
| $\mathrm{C}_{3}{ }^{2}$ | 0.4685 (4) | 0.9367(4) | 0.2660(6) | 7.1(2) |
| $\mathrm{C}_{3}^{\prime}$ | 0.3773(3) | 0.8677(3) | 0.4745(5) | 3.7(1) |
| $\mathrm{O}_{3}$ | $0.3301(2)$ | 0.8957(2) | 0.4154(3) | 4.52(8) |
| $\mathrm{N}_{4}$ | 0.3773(2) | 0.8680(2) | 0.5928(4) | 3.71 (8) |
| $\mathrm{C}_{4}^{\alpha}$ | 0.3225(2) | 0.9041 (3) | $0.6610(5)$ | 3.8(1) |
| $\mathrm{C}_{4}^{\beta 1}$ | $0.3216(3)$ | 0.9827(3) | $0.6315(7)$ | 5.6(1) |
| $\mathrm{C}_{4}^{\beta}$ | 0.3397 (3) | 0.8990(3) | $0.7918(6)$ | 5.4(1) |
| $\mathrm{C}_{4}^{\gamma}$ | 0.3420(4) | $0.8287(4)$ | 0.8487(6) | 7.9(2) |
| $\mathrm{C}_{4}^{81}$ | 0.2775(5) | 0.7943(5) | 0.8784(9) | 10.2(3) |
| $\mathrm{C}_{4}^{\delta 2}$ | $0.3806(7)$ | 0.8412(9) | 0.9729(8) | 19.6(5) |
| $\mathrm{C}_{4}^{\prime}$ | 0.2512(2) | $0.8711(2)$ | 0.6322(4) | 3.32(9) |
| $\mathrm{O}_{4}$ | 0.2429(2) | 0.8147(2) | 0.5863(3) | 4.00 (7) |
| OT | $0.1994(2)$ | 0.9125(2) | 0.6684(3) | 3.76(7) |
| C(7) | $0.1247(2)$ | 0.8947(3) | $0.6512(5)$ | 3.9(1) |
| C(8) | $0.0881(3)$ | 0.9602(3) | 0.6999(5) | 4.7(1) |
| C(9) | 0.1091 (3) | 0.8867(3) | 0.5215(6) | 6.2(1) |
| C(10) | 0.1060 (3) | 0.8301(3) | $0.722617)$ | 6.0(2) |

Anisotropically refined atoms are given in the form of the isotropic equivalent displacement parameter defined as: $(4 / 3)^{*}\left[a^{2}{ }^{*} B(1,1)+b^{2 *} B(2,2)+c^{2 *} B(3,3)+a b(\cos \gamma)^{*} B(1,2)\right.$ $\left.+a c(\cos \beta)^{*} B(1,3)+b c(\cos \alpha)^{*} B(2,3)\right]$.
carrier atom and their parameters were not refined. Final $R$ and $R_{\mathrm{w}}$ values were 0.042 and 0.040 , respectively. In the final difference Fourier synthesis the maximum and minimum electronic densities were 0.32 and -0.24 e $\AA^{-3}$, respectively. Positional atomic parameters and equivalent thermal factors for non-hydrogen atoms with their standard deviations are reported in Table 1. Tables of anisotropic thermal factors, hydrogen positional and thermal parameters, bond lengths, bond angles, torsion angles and observed vs. calculated structure factors have been deposited as Supplementary Material.

## Circular Dichroism

The CD spectra were recorded on a JASCO (Tokyo, Japan) model J-600 spectropolarimeter equipped with a Haake thermostat. Cylindrical fused quartz cells of 10 and 1 mm path lengths were employed. The values are expressed in terms of $[\theta]_{\mathrm{M}}$, the total molar ellipticity ( $\mathrm{deg} \mathrm{cm}^{2} \mathrm{dmol}^{-1}$ ). MeOH (C. Erba, Rodano, Italy) was used as solvent.

## RESULTS

## Crystal-state Conformation

The X-ray diffraction structure of $p \mathrm{BrBz}-[\mathrm{D}-(\alpha \mathrm{Me})-$ $\mathrm{Leu}]_{2}$-D-( $\left.\alpha \mathrm{Me}\right) \mathrm{Val}-\mathrm{D}-(\alpha \mathrm{Me}) L e u-\mathrm{OtBu}$ is illustrated in Figure 1. Bond lengths and bond angles are in accordance with previously reported values for the geometry of para-bromobenzamido [9] and tert-butyl ester [10] groups, ( $\alpha \mathrm{Me}$ )Leu [11] and ( $\alpha \mathrm{Me}$ )Val [12] residues, and the peptide unit [13].

The molecules of the $\mathbf{N}^{\alpha}$-blocked tetrapeptide ester are folded in an (incipient) left-handed $3_{10}$-helical structure with $\phi_{1}=58.4^{\circ}, \psi_{1}=34.4^{\circ}, \phi_{2}=56.8^{\circ}$, $\psi_{2}=18.6^{\circ}, \phi_{3}=66.6^{\circ}$ and $\psi_{3}=19.5^{\circ}$ [14]. The helix is stabilized by two consecutive $1 \leftarrow 4 \mathrm{C}=\mathrm{O} \cdots \mathrm{H}-\mathrm{N}$ intramolecular H -bonds $\left(\mathrm{N}_{3} \cdots \mathrm{O}_{0}=3.21 \AA\right.$ and $\mathrm{N}_{4} \cdots \mathrm{O}_{1}=3.12 \AA$ ). The observed $\mathrm{N} \cdots \mathrm{O}$ separations are at the upper limit for such an interaction [15]. The C-terminal D-( $\alpha \mathrm{Me})$ Leu residue is semi-extended ( $\phi_{4}=61.4^{\circ}, \psi_{4}=-164.7^{\circ}$ ). All amide, peptide and ester groups ( $\omega$ torsion angles) are trans, as expected, with only the $\mathrm{pBrBz}-\mathrm{NH} \omega_{0}$ angle deviating more than $7^{\circ}$ from planarity.


Figure 1 X -ray diffraction structure of $p \mathrm{BrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{2}-\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Val}-\mathrm{D}-(\alpha \mathrm{Me})$ Leu-OtBu with numbering of the atoms. The two intramolecular H -bonds are represented by dashed lines.

The value of the $\theta$ torsion angle of the pBrBz group, giving the orientation of the aromatic ring relative to the amide plane, is in the usual range ( $20^{\circ}$ ) [9]. The distribution of the $\chi^{1}$ and $\chi^{2}$ torsion angles characterizing the side-chain conformations is: D$(\alpha \mathrm{Me}) \mathrm{Leu}^{1} \mathrm{t}\left(\mathrm{s}^{+}, s^{-}\right)$, D- $(\alpha \mathrm{Me}) \mathrm{Leu}^{2} g^{-}\left(g^{+}, t\right), \mathrm{D}-\left(\alpha \mathrm{Me}^{2}\right) \mathrm{Val}^{3}$ $t, g^{-}$, and $\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}^{4} g^{-}\left(t, g^{-}\right)[16]$.

In the mode of packing of the tetrapeptide along the $c$-axis a helical arrangement of the molecules around the fourfold screw axis is observed. The molecules in the crystal are held together by one rather weak intermolecular hydrogen bond of the N $\mathrm{H} \cdots \mathrm{O}=\mathrm{C}$ type [15] [the $\mathrm{N}-\mathrm{H}$ group of $\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}^{1}$ in one molecule is H -bonded to the $\mathrm{C}=\mathrm{O}$ group of D $(\alpha \mathrm{Me}) \mathrm{Leu}^{4}$ of a symmetry-related molecule ( $x, 1-y$, $z-1 / 4)$ around the fourfold screw axis, with an $\mathrm{N} \cdots$ O distance of $3.12 \AA$ I , and by a number of van der Waals interactions between the hydrophobic tertbutyl, isopropyl, isobutyl and phenyl groups.

## Solution Conformation

We have previously shown that the para-bromobenzamido chromophore at the N -terminus of a peptide chain is an excellent $C D$ probe for the assignment of the screw sense of $3_{10}$-helical peptides, irrespective of the $\mathrm{C}^{\alpha}$-configuration of the constituent $\alpha$-amino acids [17]. Two intense, oppositely signed bands, positive at higher wavelengths, indicate the onset of a significant population of right-handed helical structure, while a left-handed helical structure is characterized by two oppositely signed bands, negative at higher wavelengths. The cross-over point between the two components of the exciton split curve is seen in the vicinity of 240 nm , the region where the absorption maximum of the para-bromobenzamido chromophore is found [18].

In Figure 2, the CD spectrum in the $210-300 \mathrm{~nm}$ region in MeOH solution of $p \mathrm{BrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{2}-\mathrm{D}-$ ( $\alpha \mathrm{Me}$ )Val-D-( $\alpha \mathrm{Me}$ )Leu-OtBu is compared with that of $p \mathrm{BrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{4}-\mathrm{OtBu}$ [17]. It is evident that the signs of the exciton split CD curve of the host/guest tetrapeptide associated with the para-bromobenzamido chromophore are opposite to those of the host homo-tetrapeptide. In particular, the long wavelength CD band is negative for the former, whereas it is positive for the latter. These findings are strongly in favour of the conclusion that in MeOH solutions the host/guest tetrapeptide is preferentially folded in a left-handed helix, whereas the host homo-tetrapeptide helix prefers the opposite screw sense.


Figure 2 Circular dichroism spectra in the $210-300 \mathrm{~nm}$ region of $p \mathrm{BrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{2}-\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Val}-\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}-\mathrm{OtBu}$ $(\mathrm{A})$ and $\mathrm{pBrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{4}-\mathrm{OtBu}(\mathrm{B})$ in MeOH solution (peptide concentration 1 mm ). Curve (B) was adapted from [17].

## DISCUSSION

In this article we have shown that a helix screw sense reversal is undergone by peptides formed by $C^{\alpha}$ methylated, chiral amino acids. In the crystal state, as well as in methanol solution, the terminally blocked tetrapeptide $p \mathrm{BrBz}-[\mathrm{D}-(\alpha \mathrm{Me}) \mathrm{Leu}]_{2}$ - $\mathrm{D}-(\alpha \mathrm{Me})-$ Val-D-( $\alpha \mathrm{Me}$ )Leu-OtBu is folded in a left-handed $3_{10^{-}}$ helical structure. These findings contrast dramatically with those of the host homo-tetrapeptide $\mathrm{pBrBz}-$ [D- $(\alpha \mathrm{Me}) \mathrm{Leu}]_{4}-\mathrm{OtBu}$ which is known to prefer the right-handed $3{ }_{10}$-helical handedness [11, 17], despite the two compounds possessing identical N -and

C -blocking groups. These results clearly indicate that the conformational tendency of a single, internal, $\beta$ branched $\mathrm{D}-(\alpha \mathrm{Me})$ Val guest residue (left-handed helix) [4] is strong enough to overtake that of two internal $\gamma$ branched $\mathrm{D}-(\alpha \mathrm{Me}) L e u$ residues (right-handed helix) [4]. Solution and crystal-state evidence for the coexistence of opposite screw senses in $3_{10}$-helical oligopeptides built-up of $\mathrm{C}^{\alpha}$-methylated amino acids has been reported [19, 20].

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[^0]:    Address for correspondence: Prof. Claudio Toniolo, Department of Organic Chemistry, University of Padova, via Marzolo 1, 35131 Padova, Italy.
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